MATERIALS : METALS	S AND NON-METALS			
<1M> 1.Give one exampl	le of each: metals and n	on-metals.		
2.Name the metal,	, which is the best cond	uctor of heat an	d electric	city.
3.Name the prope	rty by which metals can	be drawn into	thin wire	s. <
4.Name the gas pr	oduced, when metals re	eacts with acids		
5.What is the colo	r of the copper sulphate	e solution?		
6.State the nature	of oxides of non-metal	S.		
7.Which metal is s	tored in kerosene?			
8.Name the prope	rty of the metal by which	ch it can be drav	vn into th	nin sheets.<
9.What happens w	hen sulphur reacts with	n oxygen?		
10.Which non-met	cal catches fire, if expos	ed to air?		
11.Name the gas t	hat burns with a POP so	ound.		
12.What are Displa	acement reactions?			
13.Give one use of	non-metal in our daily	life.		
14.What are meta	lloids?			
15.Which metal is	use to wrap food items	?		
16. Which among	the following is a non-r	metal:		
(A) Copper.	(B) Zinc.	(C) Iron.	(D) Wo	od.
17.Which among	the following is a meta	1?		
(A) Zinc.	(B) Wood.	(C) Rubber.	(D) Plas	stic.
18.Which among	the following is the bes	st conductor of e	electricity	/?
(A) Silver.	(B) Iron.	(C) Aluminum (D) Copp		(D) Copper.
19.All	are good conductors	s of heat and ele	ectricity.	
	(B) Non-metals.			(D) Gases.

20. Which metal car	n be easily cut v	with a knife?			
(A) Iron.	(B) Aluminum	n. (C) Sc	odium.	(D) Zinc.	
21.Which among th	ne following sho	ows the correc	t meaning of m	alleable for metals?	
(A) They are sonore	ous.	(B) They can	(B) They can be drawn into sheets.		
(C) They can be drawn into wires.		(D) None of t	hese.		
22.Which among th	ne following sho	ows the correc	t meaning of du	uctility for metals?	
(A) They can be dra	awn into wires.	(B) Th	ney can be drav	vn into.	
(C) They are shiny.		(D) None of these.			
23.Non-metal used	in purification	of water is:			
(A) Nitrogen.	(B) Oxygen.	(C) Sulphur.	(D) Chlorine.		
24.Which metal is v	widely used in r	naking jewelle	ry?		
(A) Iron.	(B) Zinc.	(C) G	old.	(D) Copper.	
25.Zinc displaces co	opper from its s	alt solution is	an example of:		
(A) Decomposition	Reaction.		(B) Combinat	tion Reaction.	
(C) Displacement Reaction.			(D) None of these.		
26.Which metal am	ong the follow	ing is used in w	rapping the fo	od materials?	
(A) Aluminum.	(B) So	dium.	(C) Iron.	(D) None of these.	
27.Which gas is pro	duced when m	etals react wit	h sulphuric acio	d?	
(A) Oxygen.	(B) Hydrogen	. (C) Ni	trogen.	(D) None of these.	
28.Which element	catches fire if e	xposed to air?			
(A) Sulphur.	(B) Phosphor	ous.	(C) Sodium.	(D) None of these.	
29.Metals react wit	:h sodium hydro	oxide to produ	cega	s.	
(A) Hydrogen.	(B) Ox	kygen. (C) Ni	trogen. (D) N	one of these.	

30. Oxides of non-me	etais arein n	ature.			
(A) Acidic.	(B) Basic.	(C) Amphoteric.	(D) None of these.		
31. Which among the following is kept stored in water?					
(A) Iron.	(B) Zinc.	(C) Copper.	(D) Phosphorus.		
32.Sodium is kept st	ored in				
(A) Acid.	(B) Base.	(C) Water.	(D) Kerosene.		
33.What is the color	of the copper sulphat	e solution?			
(A) Green.	(B) Red.	(C) Yellow.	(D) Blue.		
34.The smallest unit	of an element is called	d A			
	(B) Molecule.		(D) None of these.		
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35.There are not more thannaturally occurring elements.					
(A) 106.	(B) 89. (C) 96.	(D) 92.			
36.The materials which allows electric current to pass through them are called as					
	(B) Insulators. (C) Ser	-			
(A) conductors.	(b) modiators. (c) ser	in conductors. (b) No	ne of the above.		
37. Which metal is used for decorating sweets?					
(A) Copper.	(B) Silver.	(C) Gold. (D) No	ne of these.		
38.Metals do not easily react with other substances are called:					
(A) Alloys.	(B) Noble.	(C) Halogens.	(D) Bases.		
39. Which metal is not used for making utensils?					
(A) Iron.	(B) Copper.	(C) Aluminium.	(D) Zinc.		
40. Which non-metal is used by plants to manufacture proteins?					
(A) Carbon.	(B) Oxygen.	(C) Nitrogen.	(D) Sulphur.		

41. Which crystalline	torm of carbon is used	d for cutting rocks?			
(A) Diamond.	(B) Graphite	(C) Lampblack.	(D) Charcoal.		
42.Which of the follo	owing is the best cond	uctor of electricity?			
(A) Silver.	(B) Copper.	(C) Aluminium.	(D) Iron.		
43.Which non-meta	l is used in manufactur	e of pesticides, like Ga	ımmexane?		
(A) Iodine.	(B) Bromine.	(C) Chlorine.	(D) Fluorine.		
44.Which non-meta	l is used in the manufa	cture of gunpowder?			
(A) Sulphur.	(B) Chlorine.	(C) lodine.	(D) Carbon.		
45. Which non-metal is used in manufacture of ammonia and urea?					
(A) Nitrogen.	(B) Sulphur.	(C) Chlorine.	(D) Oxygen.		
46.Which of the follo	owing is a liquid at ord	inary temperatures?			
(A) Carbon.	(B) Phosphorus.	(C) Oxygen.	(D) Bromine.		
47.Which of the follo	owing is/are solid non-	metals?			
(A) Carbon.	(B) Sulphur.	(C) Phosphorus.	(D) All of these.		
48.Which of the follo	owing gases is used in	advertisement signs?			
(A) Nitrogen.	(B) Oxygen.	(C) Neon.	(D) Hydrogen.		
49.Metals can be be	aten or rolled into she	ets. This property is ca	ılled:		
(A) Metallic clink.		(C) Tensile.			
50.Metals, when str	uck, produces a sound	called:			
(A) Metallic clink.	(B) Malleable.	(C) Tensile.	(D) Lustre.		
51.The solution of cl	nlorine with water is	in nature			
(A) Basic.	(B) Acidic.	(C) Neutral.	(D) None of these.		

52. Which non-metal is used to burn rocket fuel?

- (A) Nitrogen.
- (B) Sulphur.
- (C) Chlorine..
- (D) Oxygen.

<2M>

53. What happens when sulphur di- oxide reacts with water? Give the chemical reaction involved.

54. Why lemon pickle cannot be stored in an aluminium foil?

55. Write two important properties of metals.

56. Why copper cannot displace zinc from zinc sulphate solution?

57. Why immersion rods for heating liquids are made up of metallic substances?

58. What happens when iron nails are dipped in water in a test tube for a week?<

<3M>

59. What happens when iron reacts with oxygen and water? Give the chemical reaction involved.

60. What happens when copper vessel is exposed to moist air for a long time? Give the chemical reaction that takes place.

61. Why gold is preferred in making jewellery?

62. What happens when dilute sulphuric acid is poured on a zinc plate? Write the chemical reaction that takes place.

63. What happens when magnesium ribbon is burnt in air? <

64. Why metals are used in making aeroplanes, bridges, satellites etc.? <

<5M>

65. Complete the following chemical reactions.

66. What will happen when ash of magnesium is dissolved in water? Is the solution acidic or basic? What effect does litmus show in case of oxides of metals?

67.Explain the following terms. (i) Malleability (ii) Ductility (iii) Sonorous (iv) Lustrous (v) Metalloids.