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CHAPTER 15 POLYMERS WORK SHEETS

CONCEPTS:

- 1. Classification of Polymers
- 2. Uses of Polymers
- 3. Preparation of Polymers

:	1. Classification Of Polymers					
CARD :01		CARI	CARD :02			
NO	QUESTIONS NO QUESTIONS					
1	Odd One Out	П	Match the following			
1	Buna-s,Buna-N,Neoprene,Polyesters(Mol.Forces)		Α	В		
2	Protein, Starch, Cellulose, Nylon-6 (Sourse of formation)	1	Co-polymers	Bakelite		
3	Polythene,Teflone,Poly styrene,Nylon-6(Mode of	2	Homo-Polymers	Teflon		
	polymerization)					
4	Polythene, Polystyrene, Polyvinyls,	3	Thermolplastics	Buna-S		
	bakelite(Thermoplastic polymers)					
		4	Thermosettings	PVC		

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	CARD:03		CARD:04	
Ш	Fill in the Blanks	IV	Match the following	
1	Linear polymers:PVC; Cross linked polymers:		A	В
2	Fibres:Polyesters ; Thermosetting:	1	HDPE	Condensation polymers
3	Addition polymers:; Condensation Polymers:Terylene	2	PHBV	Isoprene
4	Poly amides:Nylon-66; Polyesters:	3	Natural rubber	Bio degradable
		4	Poly amides	Ziegler- Natta Catalyst
	CARD:05		CARD:06	() Y
V	True or False	VI	True or False	
1	Buna-N is an addition polymers.	1	PHBV is a bio-degradable polymer	
2	Polythene is a fibre.	2	Benzoyl peroxide is used as an initiator in polymerisation	
3	Orlon is a homopolymer	3	Chlorine is used for vulcanization of rubber.	
4	Melamine-formaldehyde is not a condensation polymer.	4	Fibres are having high t	ensile strength

2.	Uses of Polymers						
	CARD:01			CARD :02			
I	Match the following)	Match the following			
	Α	В	/	Α	В		
1	Teflon	Unbreakable Crockery	1	BUNA-S	Specialty Packaging		
2	Bakelite	Oil Seals, tank lining	2	Polyethene	Rain coats, hand bags		
3	BUNA-N	Electrical switches	3	PVC	Insulators		
4	Melamine- formaldehyde	Non-stick utensils	4	PHBV	Automobile tyres.		
	R						
	CARD:03			CARD:04			
I	Fill in the blanks		II	Fill in the blanks			
1	is used in glass reinforcing material in safety helmets.		1	Heating a mixture of raw rubber with sulphur and appropriate additive is called			
2	is used for making auto tyres,floortiles,cable insulation,etc.		2	5 % Sulphur is used as a cross linking agent in the manufacture of			
3	is used for making combs,phonograph		3	is used for the making of Paints and lacquers.			

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	records,electrical switches,etc						
4	is used for making co	nveyor belts,gaskets,etc.	4		is used for making ropes,toys,pipe	es ,etc.	
	CARD:05			CARD:0	06		
V	True of False		VI	True or	False	4	
1	Bakelite is used in the manufactups.	cture of unbreakable	1	Polystyrene is a copolymer as well as addition polymer		polymers.	
2	HDPE is used for manufacturin pipes.	g buckets,dustbins &	2	HDPE is an addition as well as thermoplastic.			
3	PAN is used as a substitute for commercial fiber as Orlon/acry		3	Condensation polymerization is otherwise called Chagrowth polymer.		d Chain	
4	Nylon -6 is not used for making	g tyre cords& fabrics.	4	Novolac is used in Paints.			
III	Preparation of polymers		<u> </u>	1			
CA	ARD :01			()			
	Column A	Column B		Q)	Column C		
Name of the polymer		Monomer of the polymer		er	Uses of the polymer		
Polythene		Tetrfluoroethene CF ₂ =CF ₂			Oil seals, Tank lining		
Teflon (Polytetrfluoroethene)		Ethene CH ₂ =CH ₂			Automobile tyres and Foot wears		
Buna S		Buta-1,3-diene + Acrylonitrile CH ₂ =CH-CH=CH ₂ + CH ₂ =CH-CN			Lubricant, Insulator and making non-stick cooking ware.		
Buna N		Buta-1,3-diene + Styrene $CH_2=CH-CH=CH_2 + C_6H_5CH=CH_2$			Insulator, ,Packing material,		
CARD:	:02						
Name of the Polymer			Name of the Monomers				
Nylon 66 Melamine + Methan		al					
Nylon	Nylon 6 3-Hydroxybutanoic acid		+ 3-				
Bakelite		ino caproic acid (CH ₂) ₅ -COOH					
Bakelit	te		(CH ₂)	₅-COOH			

HOOC(CH₂)₄COOH

 $NH_2(CH_2)_6NH_2$

PHBV (biodegradable)	Caprolactum			
Nylon 2 – Nylon 6	Phenol + Methanal			
(biodegradable)	Thenor Tweethanar			
Terylene(Dacron)	Ethane-1,2-diol + Benzene-1,4-dicarboxylic acid			

CARD:03

III.Fill in the blanks

- 1. Ethene is monomer unit for the preparation of (PVC/polythene)
- 2. Hexamethylene diamine + are used to prepare Nylon 66. (Adipic acid/ Lactic acid)
- 3. Ziegler Natta catalyst is used for the synthesis of (HDP/LDP).
- 4. PHBV is an example of polymer. (Biodegradable/ Non-Biodegradable)
- 5. Dacron/terylene are the best example of ______ (Polyamides/Polyesters)
- 6. Phthallic acid and ethylene acid are the monomers of ______. (Glyptal/Dacron

CARD:04

Write the name of the polymer whose structure is shown below:

Hint: Bakelite/Nylon-66/Buna-S/Terylene/Neoprene

$$\begin{pmatrix}
H & H & O & O \\
I & I & I & O \\
N-(CH_2)_6-N-C(CH_2)_4-C
\end{pmatrix}_{n}$$

CARD:05

Write the name of the monomers from the polymer whose structure is given below:

$$\begin{array}{c} \left(\begin{array}{c} CH_{2}\text{-}C=CH-CH_{2} \end{array}\right)_{\mathbf{n}} & \left(\begin{array}{c} H & H & O & O \\ I & I & I & I \\ N-(CH_{2})_{6}-N-C(CH_{2})_{4}-C \end{array}\right)_{\mathbf{n}} \\ \left(\begin{array}{c} CN & CN \\ CH_{2}-CH=CH-CH_{2}-CH_{2}-CH \end{array}\right)_{\mathbf{n}} & \left(\begin{array}{c} O & O \\ O & CN \\ O & CH_{2}-CH_{2}-CH_{2}-C \end{array}\right)_{\mathbf{n}} \end{array}$$

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